



Tetrahedron Letters 41 (2000) 4861-4864

A practical conversion of natural physostigmine into the potent butyrylcholinesterase inhibitor N^1, N^8 -bisnorcymserine

Xiaoxiang Zhu,^a Nigel H. Greig,^{a,*} Harold W. Holloway,^a Noel F. Whittaker,^b Arnold Brossi^c and Qian-sheng Yu^a

^aDrug Design & Development, Laboratory of Neurosciences, Gerontology Research Center (4E02), National Institute on Aging Intramural Research Program, National Institutes of Health, 5600 Nathan Shock Dr., Baltimore, Maryland 21224-6825, USA

^bLaboratory of Bioorganic Chemistry, National Institute of Diabetes and Digestive and Kidney Disease, National Institute of Health, Bethesda, Maryland 20892, USA

School of Pharmacy, University of North Carolina at Chapel Hill, Chapel Hill, North Carolina 27599-7361, USA

Received 30 March 2000; revised 28 April 2000; accepted 2 May 2000

Abstract

A rapid novel synthetic route to the potent reversible butyrylcholinesterase inhibitor (-)- N^1 , N^8 -bisnorcymserine (1) is reported from physostigmine (2) in a 20% total yield. Details on the formation of the imino-quinone 6 obtained in the oxidation of N^1 -benzylnoresermethole (4) and its conversion into N^1 -bisnoreseroline (7) are given. As expected, the product of this synthesis, (1), had identical biological activity to the same agent produced by total synthesis. © 2000 Elsevier Science Ltd. All rights reserved.

Keywords: butyrylcholinesterase; Alzheimer's disease therapeutic; physostigmine; anticholinesterase; N^1, N^8 -bisnor-cymserine.

Neuroscience and molecular biology studies show that inappropriate butyrylcholinesterase (BChE) activity increases the risk and/or progression of Alzheimer's disease. $^{1-3}$ Based on this, our group initiated studies to design and synthesize novel, potent, and highly selective reversible inhibitors of BChE to provide a candidate for therapeutic development and to test the novel hypothesis that BChE inhibitors are of benefit in the treatment of Alzheimer's disease. 4 (–)- N^{1} , N^{8} -Bisnorcymserine (1) proved to be one of the most potent and interesting inhibitors of human BChE in an initial pharmacological evaluation. However, until now, synthesis of this optically pure compound could be accomplished only by tedious multiple step chemistry, which included the resolution of the required 3S enantiomer at an intermediate stage. $^{4-7}$ It was therefore desirable to develop an efficient synthetic method to rapidly gain a large quantity of this optically

0040-4039/00/\$ - see front matter \odot 2000 Elsevier Science Ltd. All rights reserved. PII: S0040-4039(00)00740-1

^{*} Corresponding author. Fax: 410-558-8278; e-mail: greign@vax.grc.nia.nih.gov

active compound for use in further pharmacological studies. We report, herein, the synthesis of (-)- N^1 , N^8 -bisnorcymserine (1) from the commercially available alkaloid, physostigmine (2).

(–)-Physostigmine (2) was treated with sodium n-butoxide in n-butanol to give eseroline (3).⁸ (–)-Eseroline (3) then was purified and isolated as its fumarate salt, and, thereafter, was converted into N^1 -benzylnoresermethole (4), according to a known procedure N^1 -benzylnoresermethole (4).

Scheme 1. (1) Ref. 7, 89%; (2) (a) CH₃I, KOH, DMSO (b) C₆H₅CH₂NH₂, total 66%

In order to prepare the desired N^1, N^8 -bisnor derivatives, selective N^8 -demethylation of compound 4 is a key step. It was reported 12 that oxidation of N^8 -methyl to N^8 -formyl, followed by hydrolysis with diluted hydrochloric acid afforded N^8 -norphysostigmine. Following the previous methods, the oxidation of N^1 -benzylnoresermethole (4) gave the N^8 -formyl compound 5^{13} in 25 and 36% yields, respectively, when pyridinium dichromate (PDC) and Collins' reagent 14 were used as oxidants in dichloromethane. This low yield is an obstacle in the efficient conversion of physostigmine (2) to the target compound (1), particularly in light of the fact that the N^8 -formyl compound 5 proved to be a mixture of geometrical isomers in a ratio of 1:3. These were clearly detected in the 1 H NMR spectrum, and showed as two close spots by TLC, but, nevertheless, were difficult to separate. 6,7

With some modification, the yield rose to 45% when the oxidation of N^1 -benzylnoresermethole (4) was undertaken with PDC in an ice bath and in the presence of a weak base, such as pyridine or sodium bicarbonate. We additionally found that a new compound, the imino-quinone $\mathbf{6}$, was simultaneously generated and we isolated it from the reaction mixture by column chromatography in a 19% yield.

The following is a detailed experimental procedure for the oxidation of N^1 -benzylnoresermethole (4): To a solution of 4 (1.54 g, 4.99 mmol) in CH_2Cl_2 (75 mL) was added NaHCO₃ (1 g). The mixture was stirred vigorously and cooled in an ice bath. Pyridinium dichromate (3.76 g, 9.99 mmol) was then added, and the mixture was stirred for 2 h. The reaction mixture was filtered and the resulting solid was washed with CH_2Cl_2 (50 mL). The combined CH_2Cl_2 solution was washed with water (3×50 mL), dried over anhydrous Na₂SO₄ and concentrated under vacuum. The residue was purified by column chromatography (silica gel) using petroleum ether:ethyl acetate (10:1 and 10:2) as the eluent to obtain starting material 4 (0.15 g) and 5 (0.65 g, 45%), then using petroleum ether:ethyl acetate (4:1) as the eluent to obtain the more polar component 6 (0.235 g, 19%).

Compound 5 then was reacted with BBr₃ in dichloromethane for one hour at room temperature and methanol was added to destroy any remaining BBr₃. The resulting solution was evaporated to give a residue that was purified by flash chromatography to afford N^1 -benzylbisnorseroline (7), with simultaneous hydrolysis of the N-formyl group. The latter was unexpected and provided a valuable one pot reaction to effect both O-demethylation and N-deformylation, resulting in compound 7 in a yield of 77% (Scheme 2).

$$H_3$$
C H_3 C

Scheme 2. (1) PDC, NaHCO₃, CH₂Cl₂; (2) BBr₃, CH₂Cl₂, CH₃OH 77%; (3) NaBH₄, THF, 85%; (4) BBr₃, CH₂Cl₂, 91%; (5) Na, anhydrous ether, 4-isopropylphenyl isocyanate, 97%; (6) H₂, Pd(OH)₂/C, 70%

In the reaction of compound **5** with BBr₃, carefully controlling the addition of methanol to destroy remaining BBr₃ (specifically, by using approximately an equivalent of methanol and maintaining the temperature at 0°C in an ice bath), followed by dilution of the reaction mixture with dichloromethane and washing the CH₂Cl₂ solution with water to remove acid resulted in *O*-demethylation alone to give compound **8**¹⁷ in 91% yield.

Compound 6 was reduced with sodium borohydride in THF to afford the useful intermediate N^1 -benzylbisnorseroline (7)¹⁶ in 85% yield.

Compound 7 is a key intermediate from which the N^1, N^8 -bisnorcymserine was readily prepared. Reaction of 4-isopropylphenyl isocyanate with compound 7 afforded N^1 -benzyl- N^8 -norcymserine (9). N^8 -Debenzylation of compound 9 was accomplished by catalytic hydrogenation using $Pd(OH)_2/C$ as a catalyst and *iso*-propanol as a solvent to give N^1, N^8 -bisnorcymserine (1). As both of the products of the oxidation of N^1 -benzylnoresermethole (4), compounds 5 and 6, were converted into N^1 -benzylbisnorseroline (7), the synthesis of N^1, N^8 -bisnorcymserine (1) from physostigmine (2) was achieved in an overall yield that totalled 20%. Chemical characterization demonstrated that compound 1, prepared as described herein, was chemically identical in every respect to the same material prepared by total synthesis. An ex vivo assay 11 then was undertaken to quantify and compare the activity of N^1, N^8 -bisnorcymserine (1), synthesized by both routes, to inhibit human BChE and AChE prepared freshly from plasma and erythrocytes, respectively. As expected, compound 1, prepared as described herein and by total synthesis, had

identical anticholinesterase activity (with IC₅₀, concentration required to inhibit 50% enzyme action, values for BChE and AChE being 1.0 and 110 nM, respectively).

References

- 1. Guillozet, A.; Smiley, J. F.; Mash, D. C.; Mesulam, M. M. Ann. Neurol. 1997, 42, 909-918.
- 2. Barber, K. L.; Mesular, M. M.; Kraft, G. A.; Klein, W. L. Proc. Soc. Neurosci. 1996, 22, 1172.
- 3. Lehman, D. J.; Johnston, C.; Smith, A. D. Hum. Mol. Genet. 1997, 6, 1933–1936.
- 4. Yu, Q. S.; Holloway, H. W.; Utsuki, T.; Brossi, A.; Greig, N. H. J. Med. Chem. 1999, 42, 1855–1861.
- 5. Pei, X. F.; Yu, Q. S.; Lu, B. Y.; Greig, N. H.; Brossi, A. Heterocycles 1996, 42, 229-236.
- 6. Yu, Q. S.; Greig, N. H.; Holloway, H. W.; Brossi, A. J. Med. Chem. 1998, 41, 2371–2379.
- 7. Yu, Q. S.; Pei, X. F.; Holloway, H. W.; Greig, N. H.; Brossi, A. J. Med. Chem. 1997, 40, 2895–2901.
- 8. Yu, Q. S.; Schönenberger, B.; Brossi, A. Heterocycles 1987, 26, 1271–1275.
- 9. Pei, X. F.; Greig, N. H.; Flippen-Anderson, J. L.; Bi, S.; Brossi, A. Helv. Chim. Acta 1994, 77, 1412–1422.
- 10. He, X. S.; Greig, N. H.; Rapoport, S.; Brossi, A. Med. Chem. Res. 1992, 2, 229-237.
- 11. Pei, X. F.; Greig, N. H.; Bi, S.; Brossi, A.; Toome, V. Med. Chem. Res. 1995, 5, 455-461.
- 12. Takano, S.; Moriya, M.; Iwabuchi, Y.; Ogasawara, K. Chem. Lett. 1990, 109-112.
- 13. Compound **5** (pale brown crystal): mp 88–92°C; $[\alpha]_D^{22} = -42.2$ (c = 0.102, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃): 1.35 (s, 1/3 3H, C3a-CH₃), 1.41 (s, 2/3 3H, C3a-CH₃), 1.91–2.04 (m, 2H, H-3), 2.40–2.75 (m, 2H, H-2), 3.74 (s, 3H, O-CH₃), 3.85–4.15 (m, 2H, CH₂Ph), 4.86 (s, 2/3H, H-8a), 5.22 (s, 1/3 H, H-8a), 6.61–6.71 (m, 2H), 8.12 (s, 2/3H, N-CHO), 8.82 (s, 1/3H, N-CHO); HRMS: calcd: 322.1681 (M⁺), found: 322.1679.
- 14. Corey, E. J.; Balanson, R. D. J. Am. Chem. Soc. 1974, 96, 6516-6517.
- 15. Compound **6** (pale yellow syrup): $[\alpha]_D^{22} = -201.4$ (c = 0.209, CH₂Cl₂); ¹H NMR (300 MHz, CDCl₃): 1.32 (s, 3H, C3a-CH₃), 1.63–1.71 (m, 1H, H-3a), 1.85–1.96 (m, 1H, H-3b), 2.20–2.31 (m, 1H, H-2a), 2.15–2.24 (m, 1H, H-2b), 3.80 (d, J = 13.4 Hz, 1H, N-CH₂Ph), 4.21 (d, J = 13.4 Hz, 1H, N-CH₂Ph), 5.02 (s, 1H, H-8a), 6.12 (d, J = 1.7 Hz, 1H, H-4), 6.54 (dd, J = 1.7, 20 Hz, 1H, H-6); ¹³C NMR (75 MHz, CDCl₃): 23.30, 39.79, 49.70, 50.90, 54.42, 101.47, 121.65, 126.91, 128.19 (2C), 128.51 (2C), 133.37, 135.59, 138.85, 160.84, 165.63, 167.87; UV (CH₂Cl₂): $\lambda_{\text{max}} = 268$ nm, $\lambda_{\text{max}} = 330$ nm; HRMS: calcd: 278.1419 (M⁺), found: 278.1421.
- 16. Compound 7 (foam): $[\alpha]_D^{22} = -224.1$ (c = 0.054, CH_2Cl_2); 1H NMR (300 MHz, $CDCl_3$): 1.39 (s, $C3a-CH_3$), 1.86–1.94 (m, 2H, H-3), 2.58–2.73 (m, 2H, H-2), 3.71–3.84 (m, 2H, CH_2Ph), 4.46 (s, 1H, H-8a), 6.40–6.53 (m, 3H, H-4, H-6 and H-7); HRMS: calcd: 280.1576 (α M⁺), found: 280.1578.
- 17. Compound **8** (syrup): ¹H NMR (300 MHz, CDCl₃): 1.40 (s, 1/3 3H, *C*3a-CH₃), 1.45 (s, 2/3 3H, *C*3a-CH₃), 1.91–2.04 (m, 2H, H-3), 2.40–2.75 (m, 2H, H-2), 3.85–4.21 (m, 2H, CH₂Ph), 4.92 (s, 2/3H, H-8a), 5.34 (s, 1/3H, H-8a), 6.61–6.71 (m, 2H), 8.20 (s, 2/3H, *N*-CHO), 8.89 (s, 1/3H, *N*-CHO).
- 18. Compound **9** (foam): $[\alpha]_D^{22} = -157.3$ (c = 0.066, CH_2Cl_2); ¹H NMR (300 MHz, $CDCl_3$): 1.25 (d, J = 6.8 Hz, 6H, (CH₃)₂CH), 1.46 (s, 3H, C3a-CH₃), 1.98–2.11 (m, 2H, H-3), 2.70–2.98 (m, 3H, H-2 and CHMe₂), 3.80–3.90 (m, 2H, CH₂Ph), 4.54 (s, 1H, H-8a), 6.50–6.90 (m, 2H); HRMS: calcd: 441.2416 (M+), found: 441.2418. Fumarate of compound **9**: white crystal, mp 137–139°C.